



## Investigation of structural effects of Bi nanoparticle addition on HTSC BSCCO-2223, through the sol-gel method

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### ABSTRACT

In this research, the influence of Bi nanoparticles addition on the high temperature superconductor BSCCO-2223 is studied. For this purpose, the samples were synthesized by the conventional sol-gel method before adding the Bi nanoparticles with  $x= 0.00, 0.03, 0.06, 0.12\text{wt}\%$  to the superconductor BSCCO-2223. The structural characterization of all the samples was done by X-ray diffraction pattern (XRD). Also, the microstructural and grain connectivity of the samples were investigated with the "Scanning Electron Microscope" (SEM). In addition, the volume fraction of all the samples was estimated by the "Material Analysis Using Diffraction" (MAUD) software. The results show that the sample with non-added Bi has the highest volume fraction of Bi-2223 phase (77.3%) compared to the samples with added Bi nanoparticles. Furthermore, the SEM photographs show that when Bi nanoparticles addition increases, the porosity and voids increase. Therefore, the experimental results indicate that structural properties and crystallographic of samples weakened by the increase of Bi nanoparticles as an additive in the high temperature superconductor BSCCO-2223.

## 1 Introduction

After the discovery of high temperature superconducting in the BSCCO system by Maeda et. al. in 1988 [1], it was found that Bi-compounds (BSCCO) with a general formula of  $(\text{Bi,Pb})_2\text{Sr}_2\text{Ca}_n\text{Cu}_n\text{O}_{10+\delta}$  ( $n= 1, 2, 3$ ), has three basic phases and variable critical temperature values: Bi-2201 phase ( $n= 1$  and  $T_C \sim 20\text{K}$ ), Bi-2212 phase ( $n= 2$  and  $T_C \sim 85\text{K}$ ) and Bi-2223 phase ( $n= 3$  and  $T_C \sim 110\text{K}$ ) [2]. In recent years, many studies have been carried out to improve the superconducting properties of BSCCO-2223 by adding different nanoparticles in order to improve the critical current density ( $J_c$ ) and the critical transition temperature ( $T_c$ ) [3-8]. However, the influence of nanoparticles on superconductors is very important from the viewpoint of practical application.

Since Bismuth compounds provide topological conductivity properties, the interest of physicists has been taken into account in the industrial applications of high temperature superconductors. Therefore, in this study, Bi nanoparticles are added to BSCCO-2223 superconductor before observing the phase formation and the microstructural of all the samples by X-ray diffraction pattern (XRD) and SEM images, respectively.

## 2 Experimental details

In this experiment, the samples were synthesized by the conventional sol-gel method [9], and the Bi nanoparticles with  $x= 0.00, 0.03, 0.06, 0.12\text{wt}\%$  were added to  $\text{Bi}_{1.7}\text{Pb}_{0.3}\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10+\delta}$ . After a precise

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stoichiometry of Bismuth nitrate  $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$  (98%), Lead nitrate  $\text{Pb}(\text{NO}_3)_2$  (99.5%), Strontium nitrate  $\text{Sr}(\text{NO}_3)_2$  (99%), Calcium nitrate  $\text{Ca}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$  (99%), Copper nitrate  $\text{Cu}(\text{NO}_3)_2 \cdot 3\text{H}_2\text{O}$  (99.5%), amount of the precursor materials were dissolved separately in deionized water were 0.5M solutions were made. Then, the Gluconic acid ( $\text{C}_6\text{H}_{12}\text{O}_7$ ) was used as the chelating agent. The mole ratio of metal nitrates to Gluconic acid was taken as 1.7: 0.3: 2: 2: 3. When used the Gluconic acid, all compounds dissolved in deionized water. Therefore, preparation time of the sol-gel method was reduced. Afterwards, the Bi nanoparticles with  $x = 0.00, 0.03, 0.06, 0.12\text{wt}\%$  was added to the sol. The sol was heated continuously at  $80^\circ\text{C}$  under stirring to form a green dried gel. Then, the gel was fired and was grained into powders. To obtain BSCCO-2223 powders, the powders were calcined at  $800\text{-}820^\circ\text{C}$  exposed to the air for 24h to remove any remaining volatile materials. After completing the process, the powders were pressed under pressure  $5\text{Kg}/\text{m}^2$  into circular pellet with a diameter of 10mm. Afterwards, the pellet was sintered between  $800\text{-}840^\circ\text{C}$  in exposure of oxygen for 48h.

### 3 Characterization

The X-ray diffraction patterns (XRD) of all the samples were identified by using the PANalytical Company X'Pert PRO MPD PW3050/60 X-ray diffractometer with  $\text{CuK}\alpha 1$  and  $\text{CuK}\alpha 2$  radiation ( $\lambda = 1.54060\text{\AA}$ ,  $\lambda = 1.54443\text{\AA}$ ) and  $\text{CuK}\beta 1$  ( $\lambda = 1.39225\text{\AA}$ ) in the range  $2\theta = 10^\circ\text{-}80^\circ$  operated at 40kV and 40mA. The volume fraction of all the samples was refined by using the "Material Analysis Using Diffraction" (MAUD) software [10]. The grain connectivity and the microstructural of the samples were investigated by the VEGA3 TESCAN "Scanning Electron Microscope" (SEM).

### 4 Results and discussion

The X-ray diffraction patterns of the synthesized BSCCO-2223 samples with  $x = 0.00, 0.03, 0.06, 0.12\text{wt}\%$  Bi nanoparticles that were refined by the MAUD software are shown in Figs. 1-4. In these pictures, the experimental data, computed data, and subtracted data have been shown with black points, red

line and blue line, respectively. Also, the volume fraction of all the samples have been listed in Table 1. The existence of Bi-2223, Bi-2212, Bi-2201,  $\text{Bi}_2\text{CuO}_4$ , Bi and CuO phases in the XRD patterns, are consistent with the powder diffraction file of "JCPDS" (JCPDS 1540770 for  $\text{Bi}_{1.6}\text{Ca}_2\text{Cu}_3\text{O}_{10}\text{Pb}_{0.4}\text{Sr}_2$  (Bi-2223), JCPDS 1540940 for  $\text{Bi}_2\text{CaCu}_2\text{O}_{9.07}\text{Sr}_2$  (Bi-2212), JCPDS 1001691 for  $\text{Bi}_{1.916}\text{CuO}_{5.482}\text{Sr}_{1.84}$  (Bi-2201), JCPDS 1006028 for  $\text{Bi}_2\text{CuO}$ , JCPDS 4002409 for Bi, JCPDS 1011194 for CuO). Furthermore, the pattern fitness can be checked for all XRD data with the MAUD software. There are several parameters for the calculation of pattern fitness in the MAUD software. The goodness of fit (S) is described by  $S = (R_{\text{wp}} / R_{\text{exp}})$ , where  $R_{\text{wp}}$  is the weighted residual error and  $R_{\text{exp}}$  is the expected error. Refinement has been continued until a convergence was reached for a value of S close to 1, which confirms the goodness of refinement. These parameters for the samples are written in Table 2.

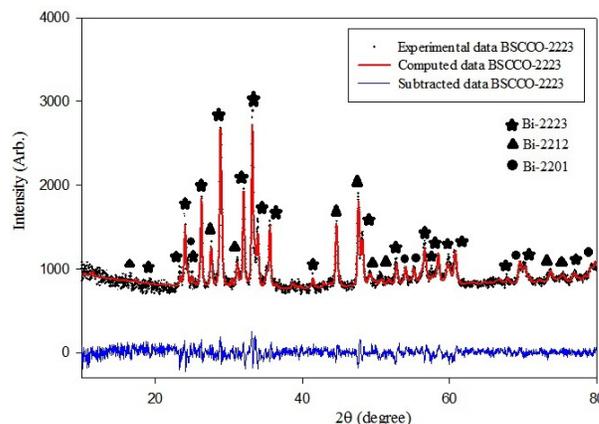


Fig 1. XRD of BSCCO-2223 (pure) sample.

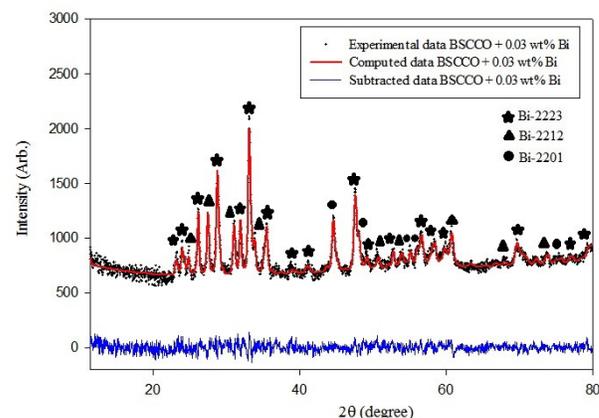


Fig 2. XRD of BSCCO-2223+ 0.03wt% Bi sample.

The results showed that the BSCCO-2223 pure sample has the majority volume fraction of Bi-2223 tetragonal structure with I4/mmm space group (77.3%) and the minority volume fraction of Bi-2212 orthorhombic structure with Amaa:cab space group (13.7%) and Bi-2201 monoclinic structure with C2/c:-b1 space group (8.9%). Also, the results indicated that the sample with x= 0.12wt% Bi nanoparticles has the lowest volume fraction of Bi-2223 phase (11.1%). However, this sample has the highest volume fraction of Bi<sub>2</sub>CuO<sub>4</sub> tetragonal structure with P4/ncc:2 space group (31.1%). In other words, when 0.12 weight percent of Bi nanoparticles was added in BSCCO-2223, it forms the Bi<sub>2</sub>CuO<sub>4</sub> tetragonal structure with P4/ncc:2 space group (31.1%), Bi triclinic structure with P1 space group (21.4%) and CuO monoclinic structure with C2/c:b1 space group (2.3%). Thus, the volume fraction of main Bi-2223 phase decreased and the volume fraction of Bi-2212 and Bi-2201 and unwanted phases increased.

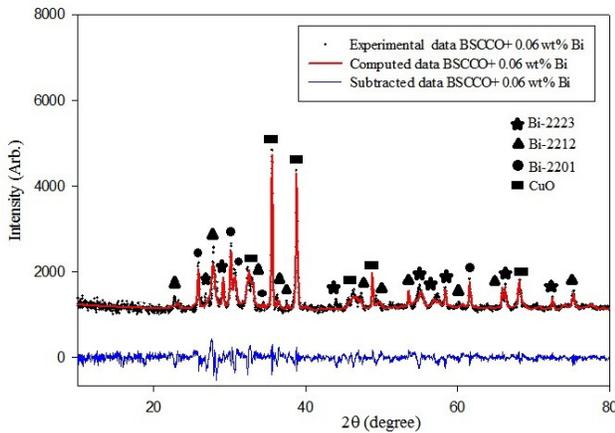


Fig 3. XRD of BSCCO-2223+ 0.06wt% Bi sample.

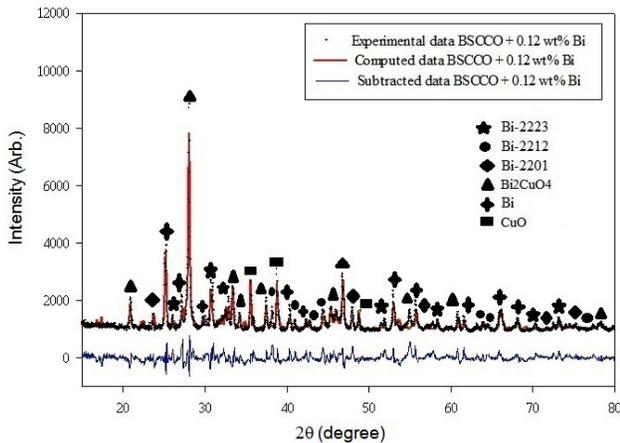


Fig 4. XRD of BSCCO-2223+ 0.12wt% Bi sample.

Table 1. The volume fraction of samples.

Composition	Bi-2223 (%)	Bi-2212 (%)	Bi-2201 (%)	Bi <sub>2</sub> CuO <sub>4</sub> (%)	Bi (%)	CuO (%)
BSCCO-2223 (pure)	77.3 (68)	13.7 (18)	8.9 (13)	–	–	–
BSCCO+ 0.03wt% Bi	65.4 (99)	33.6 (92)	0.8 (08)	–	–	–
BSCCO+ 0.06wt% Bi	25.8 (05)	26.3 (17)	12.9 (25)	–	–	34.9 (53)
BSCCO+ 0.12wt% Bi	11.1 (86)	25.6 (92)	8.1 (68)	31.1 (54)	21.4 (9)	2.3 (07)

Table 2. The calculation of pattern fitness for samples.

Composition	R <sub>wp</sub> (%)	R <sub>b</sub> (%)	R <sub>exp</sub> (%)	S
BSCCO-2223 (pure)	6.3(56)	4.9(28)	3.2(62)	1.9(4)
BSCCO+0.03wt% Bi	4.6(36)	3.5(94)	3.5(17)	1.3(1)
BSCCO+0.06wt% Bi	6.4(17)	4.6(13)	2.7(73)	2.3(1)
BSCCO+0.12wt% Bi	10.1(39)	7.2(03)	2.8(55)	2.5(5)

Electrical measurements of pure specimens and 0.03%wt are given in Figs. 5 and 6. The onset critical temperature T<sub>c</sub> (onset), the offset critical temperature T<sub>c</sub> (offset), the middle critical temperature T<sub>c</sub> (mid), and the zero electrical resistivity temperature T<sub>c</sub> (R=0) of the samples are shown in pictures.

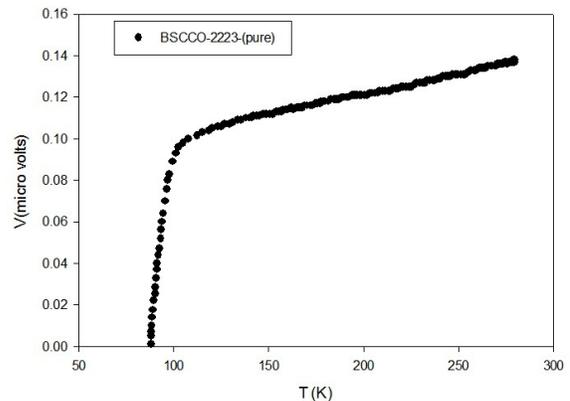


Fig 5. Electrical measurement vs. T for BSCCO-2223 (pure).

It could be readily noticed that the electrical resistivity of samples shows metallic behavior (a linear variation of resistivity with respect to temperature) in the normal state. Another feature represented by the figures is that the pure BSCCO-2223 sample has sharp superconducting transition with  $T_c$  (onset)= 104.8K,  $T_c$  (offset)= 88.2K and  $T_c$  (R=0) = 88.2K. However, the added sample has two stages of superconducting transition, i)  $T_c$  (onset)= 102.8K for Bi 2223 and ii)  $T_c$  (onset)= 85.9K and  $T_c$ (offset)= 83.6K for Bi 2212. This implies with additive Bi nanoparticles where the transition temperature of superconductivity decreases. The reason may be due to the difference between structural phases characteristic and their coupling.

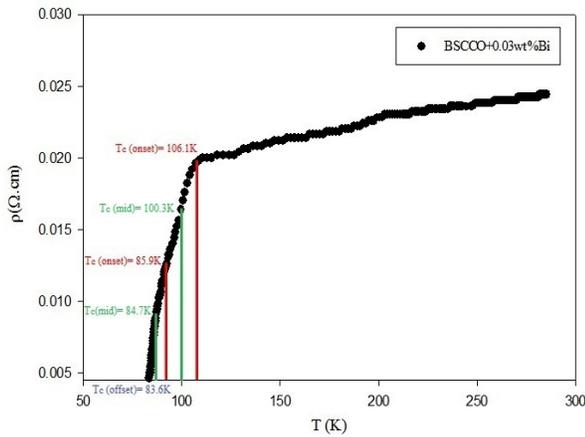


Fig 6. Electrical measurement  $\rho(T)$  for BSCCO+0.03wt% Bi.

In addition, microstructure and the grain connectivity of the samples are observed by SEM images. The SEM images of the samples are shown in Figs. 7-8.

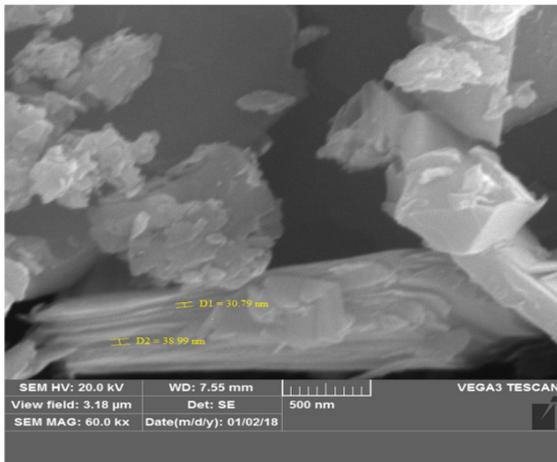


Fig 7. SEM image of synthesized BSCCO-2223 (pure).

The SEM measurements indicated that when Bi nanoparticles addition increases, the porosity and voids decrease up to 0.03 before increasing. On the other hand, the increase of porosity and voids may be related to the formation of Bi-2212 phase, Bi-2201 phase, and other secondary phases.

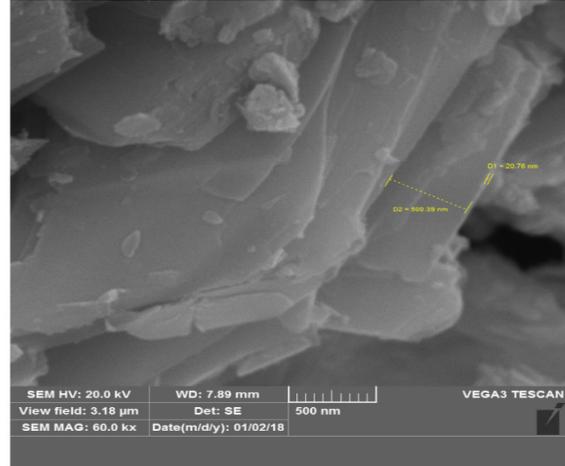


Fig 8. SEM image of synthesized BSCCO+0.03wt% Bi.

## 5 Conclusions

In this research, the structural properties and microstructure of Bi nanoparticles ( $x = 0.00, 0.03, 0.06, 0.12\text{wt}\%$ ) addition in the BSCCO-2223 superconductor were investigated. For this purpose, the samples were prepared by the conventional sol-gel method. The X-ray diffraction patterns were refined with the MAUD software. Analysis of the X-ray diffraction patterns showed that the BSCCO-2223 pure sample has the highest volume fraction of Bi-2223 phase (77.3%) compared to added samples with Bi nanoparticles. In addition, the SEM images indicated that when Bi nanoparticles addition increases, the porosity and voids increase. Electrical measurements show that with increasing Bi additive, the transition temperature of superconductivity decreases. This may be due to the formation of phases 2212 and 2201 and the unwanted phases in competition with phase 2223. Therefore, the results show that the structural properties and grain connectivity of the samples are weakened by adding Bi nanoparticles in the high temperature superconductor BSCCO-2223.

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